**Lewis diagrams (Level 2) examiners tips: Read these please!**

• draw clear diagrams, not undecipherable scribbles

• use sensibly sized dots, not dots that look like a speck of dust

• in past NCEA Exams both dots and crosses have been accepted as representing electrons

• a line or lines can be used to represent a single bond or double/triple bonds between atoms

• remember to draw in the valence electrons

• remember to draw in the lone pair(s) of electrons too

**Also…”don’t be daft”**

have we mentioned the non-bonding electrons (both valence and lone pairs)? you absolutely MUST add them in

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